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Cross multiply $X = \frac{2.5 \text{ mols}}{5 \text{ liters}}$ Level 3 Given grams instead of moles and liters of solution Determine the molarity when 117g of NaCl are dissolved to make 0.500 liters of solution. In analytical chemistry the following type of solution is used to help identify or determine the concentration of an unknown substance.

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ChemCon Chemistry constants such as R or Avogadro s number as well as equations like the combined gas law or Gay Lussac s law

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