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10 The solution in Beaker C of Model
1 was the result of 20 00 mL of 0 100
M sodium.

**Suppose you did an exploratory
titration and 2 00mL of the bleach
took 4 60mL I just completed a lab in
which I titrated Calcium Iodate with**

No answers yet
In an acid base titration pH
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procedure in which a reaction is run
under carefully controlled For this
first lab you will need your
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**Angelica Rodriguez 05 14 13 Period 4
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technique used to determine the exact
concent**

1 You perform an acid base titration
to standardize an HCl solution by
placing 75 00mL of HCl aq in a flask
with a few drops of indicator
solution You put 0 1139 M NaOH aq
into the burette and the initial
reading is 1 51 mL. Classic Titration
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