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1 Chapter 1 Review of Chemical Bonding 2 1 1 Valence bond theory Lewis dot structures the octet rule formal charge resonance and the isoelectronic principle 3 1 2 The shapes of molecules VSEPR theory and orbital hybridization

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A chemical bond is the result of an attraction between atoms or ions The types of bonds that a molecule contains will determine its physical properties such as melting point hardness electrical and thermal conductivity and solubility

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In which of the following compounds does the bond between the central atom and chlorine have the greatest ionic character

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